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THE EFFECT OF NON\_RIGIDITY ON THE LINE STRENGTES

OF VIBRATIONAL-ROTATIONAL TRANSITIONS\*

Harry C. Allen, Jr. \*\*

Mallinckrodt Chemical Laboratory Harvard University, Cambridge, Mass.

# Abstract

The effect of the difference in asymmetry of the ground and excited states of a vibrational band on the line strengths of the vibrational-rotational transitions has been investigated. It is found that for sufficiently large  $\Delta\mathcal{X}$  ( $\sim$ .1-.2) in A and C type bands there is an enhancement of the line strengths of either the R or P branch, depending on whether  $\mathcal{X}$  of the ground state is greater or less than  $\mathcal{X}$  of the excited state. In B type bands certain sub-bands are enhanced, but in general no branch of the band is enhanced as a whole.

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<sup>\*\*</sup>Atomic Energy Commission Postdoctoral Fellow.

# Introduction

In the early analyses of the rotational fine structure of vibrational bands of asymmetric rotors, the intensities of the fine structure components were estimated from the intensities of the transitions in the closer symmetric limit. Cross, Hainer and King have shown that such a procedure is dangerous.

P. C. Cross, R. M. Hainer, and G. W. King, J. Chem. Phys. 12, 210 (1944). CEK have published a very useful table of line strengths for the rigid asymmetric rotor. It is well known, however, that molecules are not rigid rotors. In fact, in those cases where complete analyses of the rotational-fine structure have been made, it has been necessary to use effective moments of inertia for each vibrational state. This difference of asymmetry in the two vibrational states, the asymmetry being measured by  $\mathcal{X} = (2b-a-c)/(a+c)$ , can have marked effect on

 $<sup>^2</sup>$  G. W. King, R. M. Hainer, and P. C. Cross, J. Chem. Phys. 11, 27 (1943). the line strengths of various transitions if  $\Delta \mathcal{H}$  between the two vibrational states is large. If  $\Delta \mathcal{H}$  is small, then the line strengths calculated for the rigid approximation are sufficient. However, for some of the higher combination bands of light molecules such as water and hydrogen sulfide  $\Delta \mathcal{H}$  can be as large as .1 or .2 and in free radical spectra even larger changes can occur.

<sup>&</sup>lt;sup>3</sup> This was kindly pointed out by Dr. Hersberg at the "Symposium on Melecular Structure and Spectroscopy," Columbus, Ohio, 1953.

In these cases the intensity pattern calculated for the rigid approximation breaks down.

## • Theory

The problem of the asymmetric rotor has been treated in great detail. 1,2

In setting up the energy matrix, Wang combinations of symmetric rotor wave

In setting up the energy matrix, Wang combinations of symmetric rotor wave functions must be used. Then an orthonormal transformation can be found which diagonalizes the energy, i.e., such that

$$E(\mathcal{K}) = T' X' E X T, \tag{1}$$

where X is defined as reference 2, equation 29, and serves to give the Wang combinations of the symmetric rotor wave functions, and T is the transformation which diagonalizes the energy matrix.

The line strengths are obtained by squaring the matrix elements of the direction cosine matrix of the asymmetric rotor. If the direction cosine matrix is also set up in terms of symmetric rotor wave functions, then the asymmetric rotor direction cosine matrix is obtained by

$$\mathbf{\bar{a}}_{\mathrm{Fg}}^{\mathrm{A}} = \mathbf{T}_{1}^{\mathrm{i}} \times^{\mathrm{i}} \mathbf{\bar{a}} \times \mathbf{T}_{2}^{\mathrm{i}}, \tag{2}$$

where  $T_1$  and  $T_2$  are the transformations which diagonalize the ground and excited states respectively. The case where  $T_1 = T_2 = T$  was considered by Cross, Hainer and King. In this paper the case will be considered where  $T_1 \neq T_2$ .

The transformations for  $\mathcal{H}$  = .5 and  $\mathcal{H}$  = 1 have been evaluated up through J = 4 and applied to the direction cosine matrix in the order

$$\bar{\mathbf{n}}_{Tg}^{A} = \mathbf{T} \ (.5) \ X' \ \bar{\mathbf{n}} \ X \ \mathbf{T}(1)$$
 (3)

This order assumes that  $\mathcal{X}=.5$  in the ground state and  $\mathcal{X}=1$  in the excited state. A change this large is likely to be encountered only in the case of free radical spectra, but the large change was used in order to emphasize any regularities which might appear. The resulting elements were squared and

multiplied by 10<sup>4</sup> in order to be consistent with reference 1. The transitions were then sorted into sub-tranches in order to find any regularities which occurred.

### Discussion

The regularities found are summarized in Table I. It was found necessary to consider two cases; (1)  $\mathcal R$  of ground state less than  $\mathcal R$  in the excited state, and (2)  $\mathcal R$  of ground state greater than  $\mathcal R$  in the excited state. The conclusions for case 2 are the converse of the conclusions for case 1, so only case I will be discussed.

The most striking regularities are found in the A and C type bands, i.e., the dipole moment change along the least and largest inertial axes respectively. In the A type band it is found that the P branch line strengths are enhanced, quite generally, with respect to the R branch, while in the C type bands it is the R branch which is enhanced. Certain of the Q branch sub-bands are also enhanced under these conditions.

An inspection of the table of line strengths in reference 1 shows that it is not always necessary to consider the difference of asymmetry of the two vibrational states. For an A type band, the change of line strength with  $\mathcal{H}$  is very small in the region  $-1 \leq \mathcal{H} \leq 0$ . Hence a  $\Delta \mathcal{H}$  in this region would not lead to pronounced effects. However, for  $0 \leq \mathcal{H} \leq 1$  the change in line strength with  $\mathcal{H}$  is quite steep; hence in this region of  $\mathcal{H}$  one might expect quite pronounced effects. For C type bands the pronounced effects are to be expected in the region  $-1 \leq \mathcal{H} \leq 0$ . The line strengths of Q branch transitions have a much greater  $\mathcal{H}$  dependence than either the F or R branch transitions. Such unexpected intensity patterns as can arise from these Q branch transitions could become a useful tool in the analysis of vibrational-rotational bands.

In B type bands, i.e. the dipole moment change along the intermediate inertial axis, no particular branch is generally enhanced. It is found that certain subbands in the P, Q and R branches are enhanced. It is again to be hoped that the intensity patterns found in this special case of large  $\Delta \mathcal{L}$  will become a useful analytical tool. In order to give some idea of the size of these effects, in Table II are shown the calculated line strengths for the main sub-bands for both A and B type selection rules through J = 4. No C type examples are given for the effect is very small in the  $\mathcal{L}$  region chosen, as pointed out earlier. A ground state of 0.5 and an excited state  $\mathcal{L}$  of 0.7 have been used in the calculations. The line strengths for the rigid approximations in which  $\mathcal{L} = 0.5$  and 1.0 have been included for comparison.

In many instances these effects are small, of the order of 1 or 2% for a  $\Delta\mathcal{H}$  = .2. However, for many Q branch transitions in the appropriate  $\mathcal{H}$  range these effects can be as much as 10% even at low J. In the case of free radicals

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D. A. Ramsay, Private communication.

where  $\Delta$  \( \)\( \)\) may be as high as 0.6 - 0.8 then these effects will be pronounced in practically all transitions.

 $\mathcal{X}_{G}$  = ground state

 $\mathcal{K}_{\mathbf{e}}$  = excited state

Sub-band labelling is that suggested in reference 1.

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. for the letter

Table II

A type Band

	•	80					
	ar .			<b>%</b> I .			
Transition	8 <sup>2</sup> (1)	e <sup>2</sup> (.5-7)	<b>a</b> <sup>2</sup> (.5)	Transition	<b>e</b> <sup>2</sup> (1)	5 <sup>2</sup> (.5-7)	a <sup>2</sup> (.5)
1 - 10	15000	15000	15000	10 - 11	15000	15000	19000
5 <sup>5</sup> - 5 <sup>7</sup>	25000	28223	26223	2 <sub>1</sub> - 2 <sub>2</sub>	25000	26926	2023
3 <sub>5</sub> - 3 <sub>8</sub>	35000	44966	45104	3 <sub>2</sub> - 3 <sub>5</sub>	35000	41498	45104
hh = h5	45000	67926	64494	43 - 44	45000	58949	64494
20 - 2-1	8555	8333	8555	2 <sub>-1</sub> - 2 <sub>0</sub>	8555	8555	8335
3 <sub>1</sub> - 5 <sub>0</sub>	14585	16278	16278	3 <sub>0</sub> - 3 <sub>1</sub>	14583	15514	16278
4 <sub>2</sub> - 4 <sub>3</sub>	20250	26100	26168	h <sub>1</sub> - h <sub>2</sub>	20250	23431	261.68
3 <sub>-1</sub> - 3 <sub>-2</sub>	8750	7730	7403	3 <sub>-2</sub> - 3 <sub>-1</sub>	8750	7505	7403
h <sub>0</sub> - h <sub>-1</sub>	15750	13893	13221	4 <sub>-1</sub> - 4 <sub>0</sub>	15750	12987	13221
h_2- h_3	9000	8087	7587	4-5- 4-2	9000	7616	7587
	••921			•	•°021		
2-2- 21	8353	5110	5110	21 - 2-5	8333	6405	5110
3 <sub>-1</sub> - 3 <sub>2</sub>	14985	5395	5722	3 <sub>2</sub> - 3 <sub>-1</sub>	14583	9728	5722
4 <sub>0</sub> - 4 <sub>3</sub>	20250	<b>3</b> 691	4363	43 - 40	20250	9896	4365
3 <sub>-3</sub> - 3 <sub>0</sub>	8750	7055	7055	3 <sub>0</sub> - 3 <sub>-3</sub>	8750	7819	7055
4-e- 41		10714	11214	4 <sub>1</sub> - 4 <sub>-2</sub>	15750	13952	11214
4_4- 4_1	9000	7565	7558	4 <sub>-1</sub> - 4 <sub>-4</sub>	9000	8138	7558

	** R <sub>01</sub>				ae Poī		
Transition	<b>4</b> <sup>2</sup> (1)	<b>6</b> <sup>±</sup> (.5-7)	<b>5</b> <sup>2</sup> (.5)	Transition	<b>5<sup>2</sup>(1)</b>	<b>5</b> <sup>2</sup> (.5-7)	s <sup>2</sup> (.5)
00 - 1-1	10000	19000	10000	1-1-00	10000	10000	10000
1-1- 2-2	15000	16159	16934	2-2-1-1	15000	16934	16934
2-2-3-3	25000	25351	25893	3-3-2-2	25000	25968	25893
3_3- 4_h	35000	35373	35773	4_4- 3_3	35000	35793	35 <b>773</b>
1 <sub>1</sub> - 2 <sub>0</sub>	15000	15000	15000	2 <sub>0</sub> - 1 <sub>1</sub>	15000	15000	15000
2 <sub>0</sub> - 3 <sub>-1</sub>	<b>1666</b> 7	20381	22500	3-1-20	16667	22500	22500
3,1-4,2	26250	27225	29261	4-2-3-1	<b>2</b> 6250	29554	29261
22 - 31	20000	18530	18636	3 <sub>1</sub> - 2 <sub>2</sub>	20000	19229	18636 .
3 <sub>1</sub> - 4 <sub>0</sub>	18750	25684	29055	4 <sub>0</sub> - 3 <sub>1</sub>	18750	29431	29055
33 - 42	25000	19832	20331	4 <sub>2</sub> - 3 <sub>5</sub>	25000	22214	20331
	ac R <sub>01</sub>				ao Poi	r	
10 - 2-1	15000	15000	15000	2_1-10	15000	15000	15000
2-1-3-2	25000	25445	25710	3 <sub>-2</sub> - 2 <sub>-1</sub>	25000	25710	25710
3-2-4-3	35000	35406	35758	4-3-3-2	35000	35765	35758
2 <sub>1</sub> - 3 <sub>0</sub>	16667	16667	16667	3 <sub>0</sub> - 2 <sub>1</sub>	16667	16667	16667
3 <sub>0</sub> - 4 <sub>-1</sub>	26250	<b>2</b> 7566	28252	4-1- 30	26250	28258	28252
3 <sub>2</sub> - 4 <sub>1</sub>	18750	18171	18207	41 - 32	18750	18414	18207

Table II (cont.)

	be			be				
٩ <sub>1</sub> ,٢				$\mathbf{q}_{\overline{1},1}$				
Transition	<b>5</b> <sup>2</sup> (1)	<b>8</b> <sup>2</sup> (.5-7)	<b>5<sup>2</sup>(.</b> 5)	Transition	<b>5</b> <sup>2</sup> (1)	<b>a</b> <sup>2</sup> (.5-7)	<b>5</b> <sup>2</sup> (.5)	
1-1-11	15000	15000	15000	1, -1,1	15000	15000	15000	
2 <sub>0</sub> - 2 <sub>2</sub>	25900	22912	21289	22 - 20	25000	21289	21289	
3 <sub>1</sub> - 3 <sub>3</sub>	35000	27961	23196	3 <sub>3</sub> - 3 <sub>1</sub>	35000	23041	23196	
42 - 44	45000	29930	22157	h, - hg	45000	21541	22157	
2_2-20	8535	12044	12044	20 - 2-2	8333	10422	12044	
3 <sub>-1</sub> - 3 <sub>1</sub>	14583	<b>2552</b> 9	24417	3_1-31	14585	19652	24417	
<b>h</b> <sub>0</sub> - <b>h</b> <sub>2</sub>	20250	40243	36119	ps - p0	20250	28245	36119	
3_3- 3_1	8750	11750	10583	3 <sub>-1</sub> - 5 <sub>-5</sub>	8750	9471	10583	
<sup>h</sup> 2 - <sup>h</sup> 0	15750	21904	20622	h <sub>0</sub> - h <sub>-2</sub>	15750	16498	20622	
h_h- h-2	9000	10655	10617	l <sub>k</sub> = l <sub>k</sub>	9000	9776	10544	
oo q <sub>1,</sub> ſ				ьо <sup>Q</sup> Г,1				
			•				0000	
2-1-21	8535	8333	8555	2 <sub>1</sub> - 2 <sub>-1</sub>	8333	8553	8333	
3 <sub>0</sub> - 3 <sub>2</sub>	14585	14585	13160	3 <sub>2</sub> - 3 <sub>0</sub>	14583	13160	15160	
4 <sub>1</sub> - 4 <sub>3</sub>	20250	20171	16126	43 - 41	20250	16092	16126	
3_2- 30	8750	10173	10173	3 <sub>0</sub> - 5 <sub>-2</sub>	8750	9592	10173	
4 <sub>-1</sub> - 4 <sub>1</sub>	15750	19977	18280	h <sub>1</sub> - h <sub>-1</sub>	15750	16716	18280	
A_x- A_1	9000	10515	10584	h_1 - h_5	9000	9839	10584	

Table II (cost.)

	<b>&gt;</b>				<b>bo</b>			
	B <sub>1,</sub> ;	ı			P <sub>T,1</sub>			
Premoition	<b>6</b> <sup>2</sup> (1)	<b>5</b> <sup>2</sup> (.5-7)	<b>5</b> <sup>2</sup> (.5)	Transition	<b>5</b> <sup>2</sup> (1)	<b>5<sup>2</sup>(.5-7)</b>	<b>s</b> <sup>2</sup> (.5)	
00 - 10	10000	10000	10000	10 - 00	10000	10000	10000	
1-1-2-1	15000	15000	15000	2-1-1-1	15000	15000	15000	
2-2-3-2	25000	24330	24086	3-e- 2-e	25000	24183	24086	
3 <sub>-5</sub> - 4 <sub>-5</sub>	35000	34421	34083	1-3 <sup>-3</sup> -5	35000	34114	54083	
1, - 2,	15000	15000	15000	2, - 1,	15000	15000	15000	
2 <sub>0</sub> - 3 <sub>0</sub>	16667	16667	16667	3 <sub>0</sub> - 2 <sub>0</sub>	<b>16</b> 667	16667	16667	
5-1-4-1	26250	24034	<b>23</b> 550	4-1-5-1	26250	23941	25950	
22 - 32	20000	21295	21583	3 <sub>2</sub> - 2 <sub>2</sub>	20000	20414	21383	
51 - 41	18750	19506	19563	41 - 31	18750	19182	19565	
35 - 43	25000	<b>2920</b> 5	29580	4 <sub>3</sub> - 3 <sub>5</sub>	25000	27992	29580	
	bo R <sub>1,Σ</sub>				be P <sub>1,1</sub>			
1 <sub>0</sub> - 2 <sub>2</sub>	5000	6253	7226	<b>2</b> - 10	5000	7226	7226	
2, - 3,	10000	14021	16667	3 <sub>3</sub> - 2 <sub>1</sub>	10000	16667	16667	
3g - 4k	15000	23369	27406	44 - 32	15000	27337	27406	
2-1- 51	1667	2245	2792	3 <sub>1</sub> - 2 <sub>-1</sub>	1667	2792	2792	
3 <sub>0</sub> - 4 <sub>2</sub>	3750	5675	7602	42 - 30	3750	760 <b>2</b>	7602	
<sup>5</sup> -2 <sup>- 4</sup> 0	1250	1185	1537	40 - 3-2	1250	1705	1537	